

STUDY PROGRAM TO IMPROVE FUEL CELL
PERFORMANCE BY PULSING TECHNIQUES

1ST QUARTERLY REPORT

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TABLE OF CONTENTS

	<u>Page</u>
LIST OF ILLUSTRATIONS	ii
1.0 PURPOSE	1
2.0 ABSTRACT	2
3.0 MEETINGS OR CONFERENCES	3
4.0 FACTUAL DATA	4
5.0 FUTURE WORK	7
APPENDIX	8

APPENDIX
LIST OF ILLUSTRATIONS

Figure

1. Drawing of Teflon Cell Used for Electrolyte and Gas Pulsing.
2. Drawing of Stainless Steel Electrode Holder and Electrolyte Room.
3. System for Subsonic Gas and Electrolyte Pulsing.
4. Table I Data - Polarization Curves Prior to Pulsing.
5. Oscilloscope Trace of Anode-Cathode Fluctuations at 55 Pulses/Minute.
6. Oscilloscope Trace of Anode-Cathode Fluctuations at 220 Pulses/Minute.
7. Table II Data - Polarization Curves After Electrolyte Pulsing.
8. Oscilloscope Trace of Pulsating 60-Cycle Direct Current - 50% Duty Cycle.
9. Table III Data - Polarization Curves for Electrically-Pulsed Cells.

1.0 PURPOSE

This investigation is concerned with the effect on performance of porous fuel cell electrodes by changing the gas-electrolyte interface by pulsing techniques. The interface will be disturbed by electrical and mechanical pulses at frequencies in the subsonic, sonic and ultrasonic range. Carbon (hydrophobic), metal (hydrophillic) and composite electrodes will be used in this investigation.

2.0 ABSTRACT

Equipment was designed and constructed to provide electrolyte and gas pulsing in the subsonic range (0 to 425 c. p. m.) at controlled amplitudes. Preliminary experiments on electrolyte pulsing using wetproofed carbon electrodes showed cell voltage fluctuations of approximately 60 mv at a pressure amplitude of 12 psig at moderate current densities. The voltage amplitude was independent of frequency over the range 55 to 220 c. p. m.

Data obtained thus far indicate that there is no significant polarization difference between 1/4" electrodes operating on continuous d. c. and those operating on 60-cycle interrupted d. c.

The principal investigator was abroad for almost half of this reporting period.

3.0 MEETINGS AND CONFERENCES

G. E. Evans, K. V. Kordes, and M. L. Kronenberg met with W. J. Nagel, Project Supervisor from NASA on 24 June, 1963 to discuss the scope and program of the contract. It was agreed that initial studies would involve the application of mechanical and electrical pulses that could change the position of the gas-electrolyte interface within the electrode. Three kinds of electrodes (carbon, metal and composite) would be tested at subsonic, sonic and ultrasonic frequencies. The effect of pulse intensity and shape on performance would also be studied. Temperature effects would be studied on optimized systems.

4.0 FACTUAL DATA

4.1 Test Equipment.

A drawing of the Teflon cell that was designed and built for electrolyte and gas pulsing is shown in Figure 1. The electrode size was chosen to accommodate the porous metal anodes and cathodes that are available to us. Accesses to the electrolyte room are for filling, pressure readings, and reference electrode. A drawing of the stainless steel electrode holders and gas room is shown in Figure 2. An O-ring gasket between the electrode and the cell prevents the electrolyte from penetrating around the electrode.

For electrolyte and gas pulsing in the subsonic range, a 1/8-horsepower motor with a gear box for varying speeds has been coupled with a piston and eccentric cam arrangement to provide controlled amplitudes and frequencies. Electrolyte or gas pulsations from 0 to 425 cycles per minute at amplitudes high enough to displace liquids throughout the total electrode thickness can be provided by the system which is shown in Figure 3. The pressure at the electrode surface is presently sensed with a Marsh pressure gage but a sensitive transducer coupled with a Bourdon tube will be used later for the more precise experiments.

Carbon electrodes have been set aside to provide standard carbon material for the entire testing program. An ample supply of porous metal and composite electrodes will also be available shortly.

4.2 Experimental Results.

Electrolyte pulsing experiments in the subsonic range were begun using 1/4-inch carbon electrodes. A Zn-wire reference electrode was used for these preliminary experiments. The effective area of the working electrodes was 4.9 cm². The polarization curve (prior to the pulsing experiment) is given in data form in Table I.

TABLE I

A - C	Volts		Load (ma)
	C - R	A - R	
0.980	1.383	0.404	O. C.
0.880	1.294	0.414	20
0.837	1.267	0.430	50
0.793	1.249	0.455	100
0.753	1.242	0.488	150

Note 1: A = Anode; C = Cathode; R = Reference.

Note 2: Data for Table I is plotted in Fig. 4

An oscilloscope trace of A - C (anode-cathode) fluctuations at 55 pulses per minute is shown in Figure 5, and at 220 pulses per minute in Figure 6. All traces are 2 sec./cm and represent an amplitude of 0 to 12 psi. The vertical scale is 100 mv/cm and the peak voltage corresponds to maximum displacement (maximum pressure). The voltage amplitude appears to be independent of the pulsing frequency over this limited range.

After the series summarized in Table I was run, additional electrolyte was added to the cell so that a higher amplitude would result. The increased amplitude did not produce any increase in the voltage effect, and on continued pulsing the voltage fluctuations disappeared. Upon examining the electrodes it was observed that electrolyte penetration had occurred significantly into the electrode, perhaps too deeply for electrolyte pulsing to be of any effect. The new polarization curve data (after the higher pulsing) is shown in Table II.

TABLE II

A - C	Volts		Load (ma)
	C - R	A - R	
0.980	1.460	0.470	O. C.
0.860	1.350	0.485	50
0.825	1.330	0.595	100
0.760	1.300	0.540	200

Note 1: A = Anode; C = Cathode; R = Reference

Note 2: Data for Table II is plotted in Fig. 7

Cathode performance improved while anode performance deteriorated as a result of this preliminary electrolyte pulsing. The voltage fluctuations observed in Figure 5 and 6 are encouraging, being of a higher magnitude than anticipated. However, since the results are preliminary, much more data must be obtained before any solid conclusion can be reached.

4.3 Comparison of 60-Cycle Pulsing and Direct Current.

Two cells were set up to run on nonfluctuating direct current at 10 ma/cm² and 50 ma/cm². Two other cells were set up to run at 10 ma/cm² and 50 ma/cm² at pulsed-direct current, the currents stated being average currents. The currents were checked across 1-ohm precision resistors with an electrometer, and also displayed on an oscilloscope, as shown in Figure 8.

The vertical scale is 200 mv/cm, and the sweep time was in the 10 millisecond/cm range. The curve shows current fluctuations corresponding to an average current of 50 ma/cm².

Polarization data have been taken on all four cells for 20 days thus far and the results are given in Table III. The cell voltages reported are resistance-free values.

TABLE III

No. of Days Run	Cell Voltage			
	Nonpulsating, d. c.		60-Cycle, d. c.	
	10 ma/cm ²	50 ma/cm ²	10 ma/cm ²	50 ma/cm ²
1	0.898	0.846	0.902	0.843
4	0.906	0.852	0.912	0.851
6	0.920	0.853	0.921	0.857
10	0.932	0.856	0.932	0.851
12	0.927	0.855	0.930	0.855
14	0.927	0.851	0.930	0.855
20	0.939	0.863	0.936	0.858

Note: Data for Table III is plotted in Fig. 9.

The above data show there is no significant difference in performance between continuous d. c. and 60-cycle pulsating d. c. within the 20-day period of testing.

5.0 FUTURE WORK

During the next period electrolyte pulsing experiments will be continued in the subsonic region with carbon, metal and composite electrodes at two or more pressure amplitudes. The effect of pulsing on air operation is of particular interest and this too will be studied with all three types of electrodes. Experiments will then be conducted to see if similar effects can be accomplished by moving the gas-electrolyte interface from the gas side by subsonic fluctuations in gas pressure.

During the next period we also expect to design and construct equipment suitable for extending the above measurements to the sonic range. Pulsating-current experiments with carbon electrodes will continue and similar tests will be set up with metal and composite electrodes.

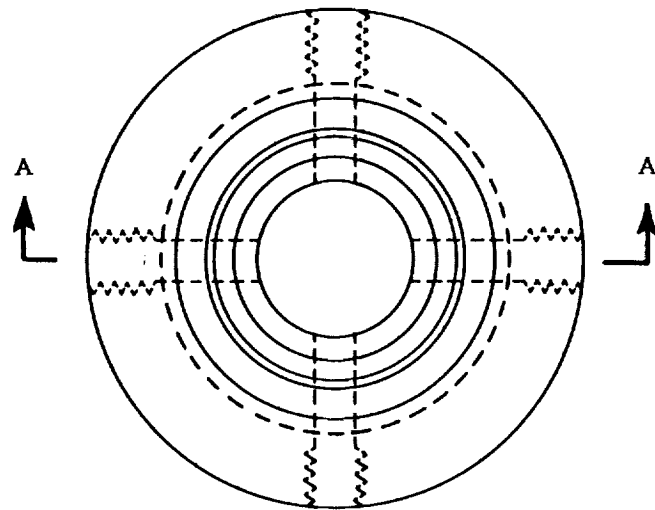
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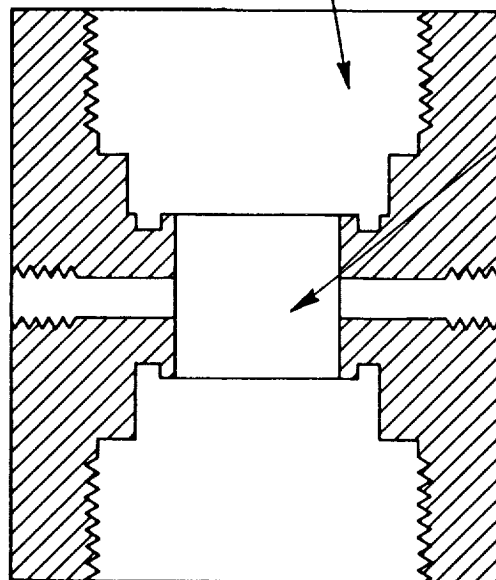
APPENDIX

FIGURE 1
DRAWING OF TEFLON CELL
USED FOR ELECTROLYTE AND GAS PULSING



Front View

Electrode and Gas Room

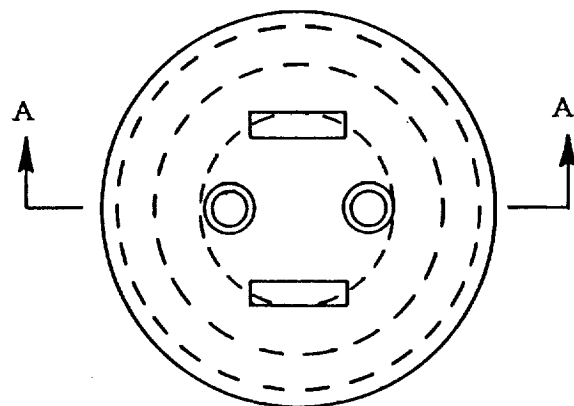


Electrolyte Room

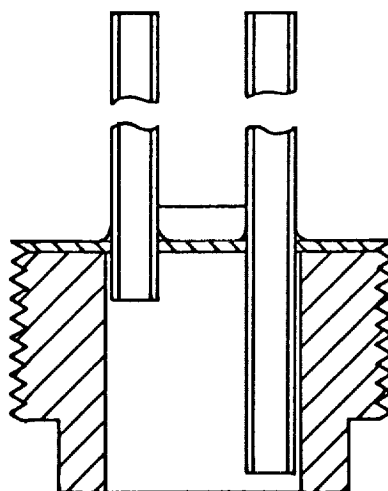
Section AA

FIGURE 2

DRAWING OF STAINLESS STEEL
ELECTRODE HOLDER AND ELECTROLYTE ROOM



Front View



Section AA

FIGURE 3

SYSTEM FOR SUBSONIC GAS AND ELECTROLYTE PULSING

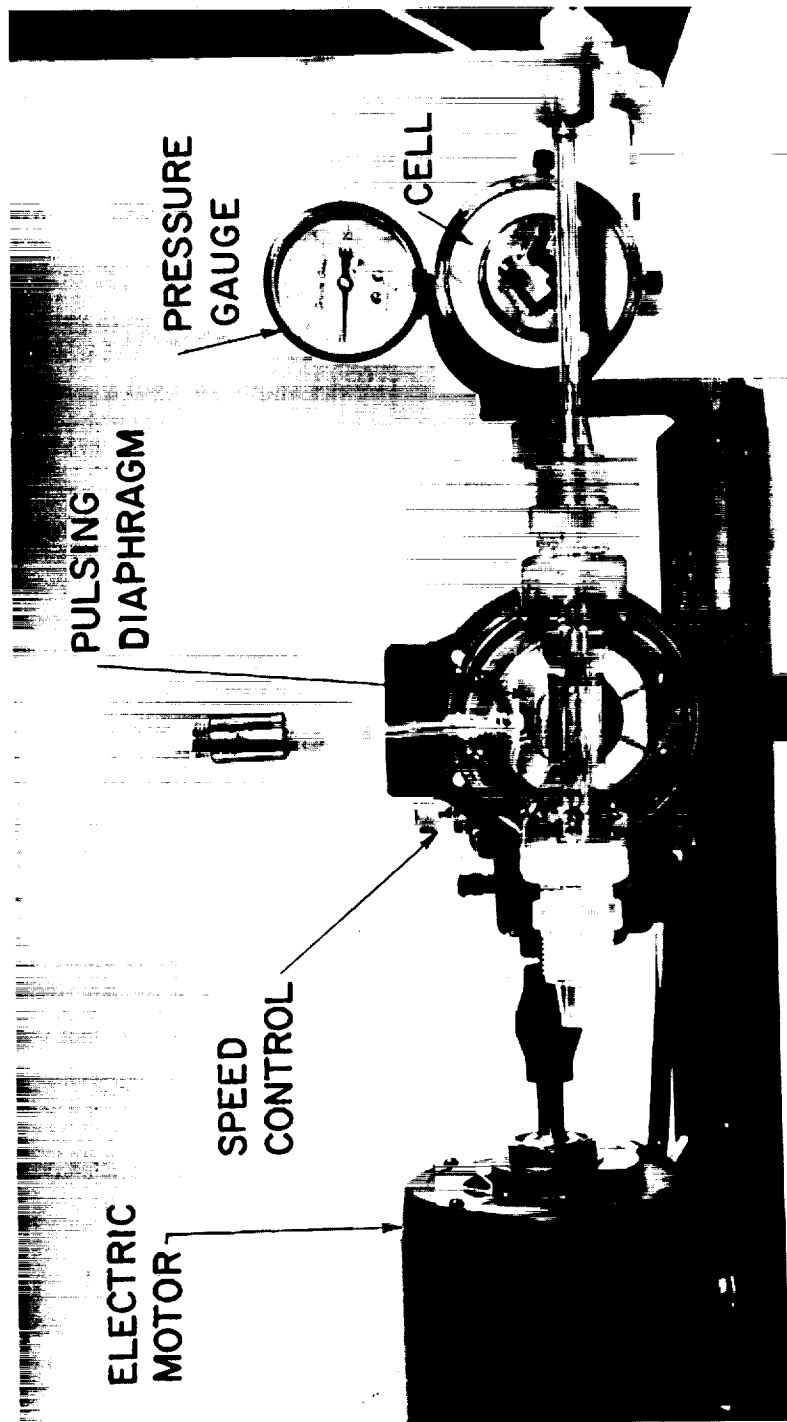
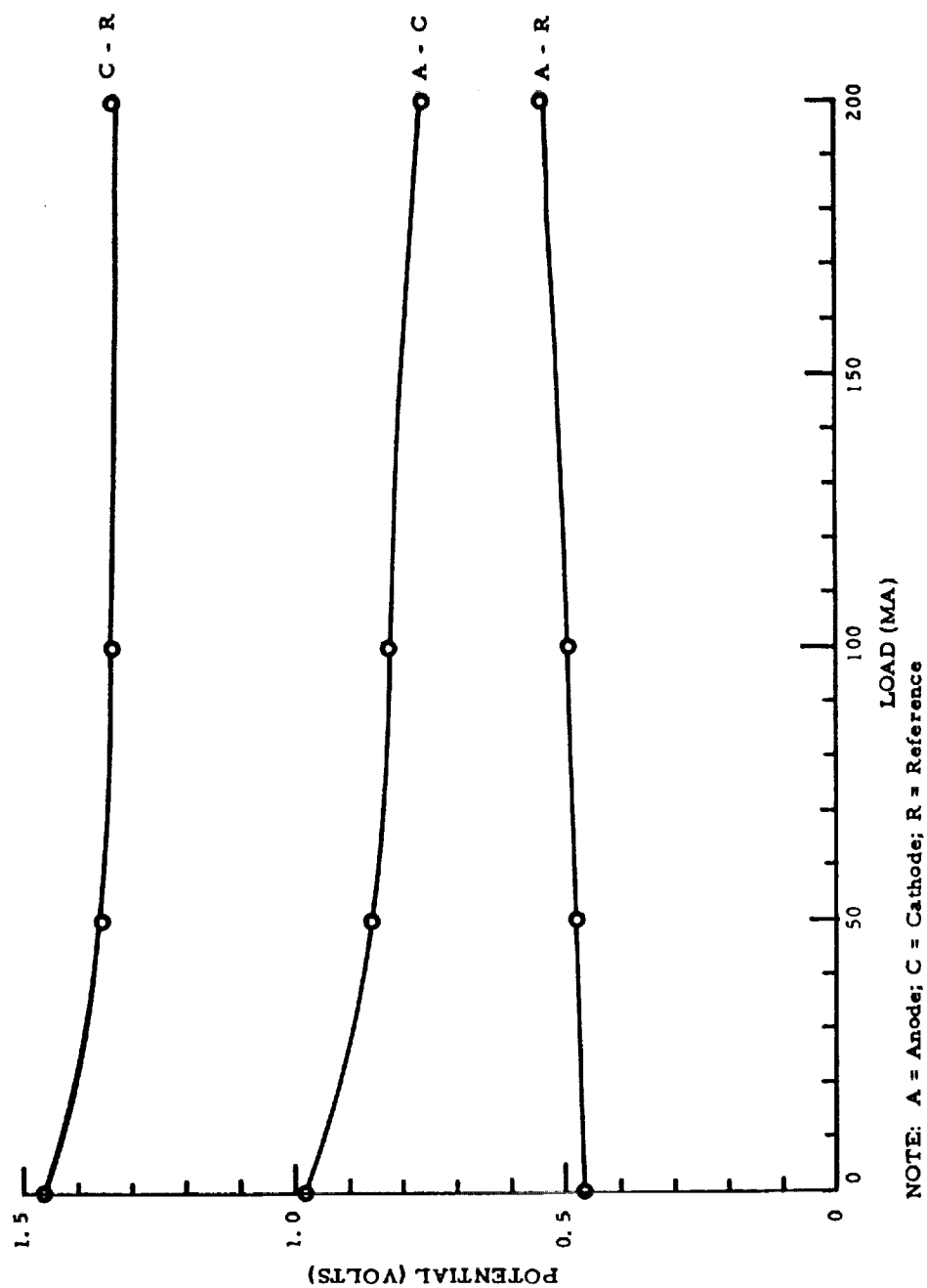
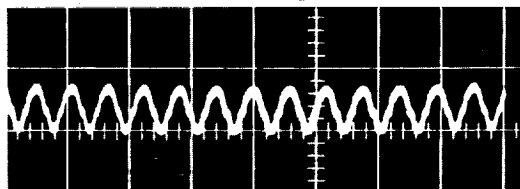


FIGURE 4

TABLE I DATA - POLARIZATION CURVES PRIOR TO PULSING



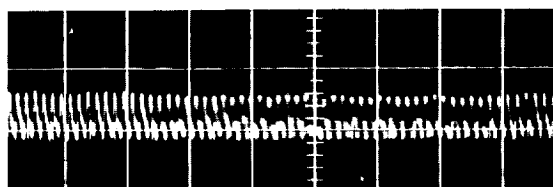
Potential
(100 mv/cm)



Time (2 sec/cm)

Fig. 5 Oscilloscope Trace of Anode-Cathode
Fluctuations at 55 Pulses/Min.

Potential
(100 mv/cm)

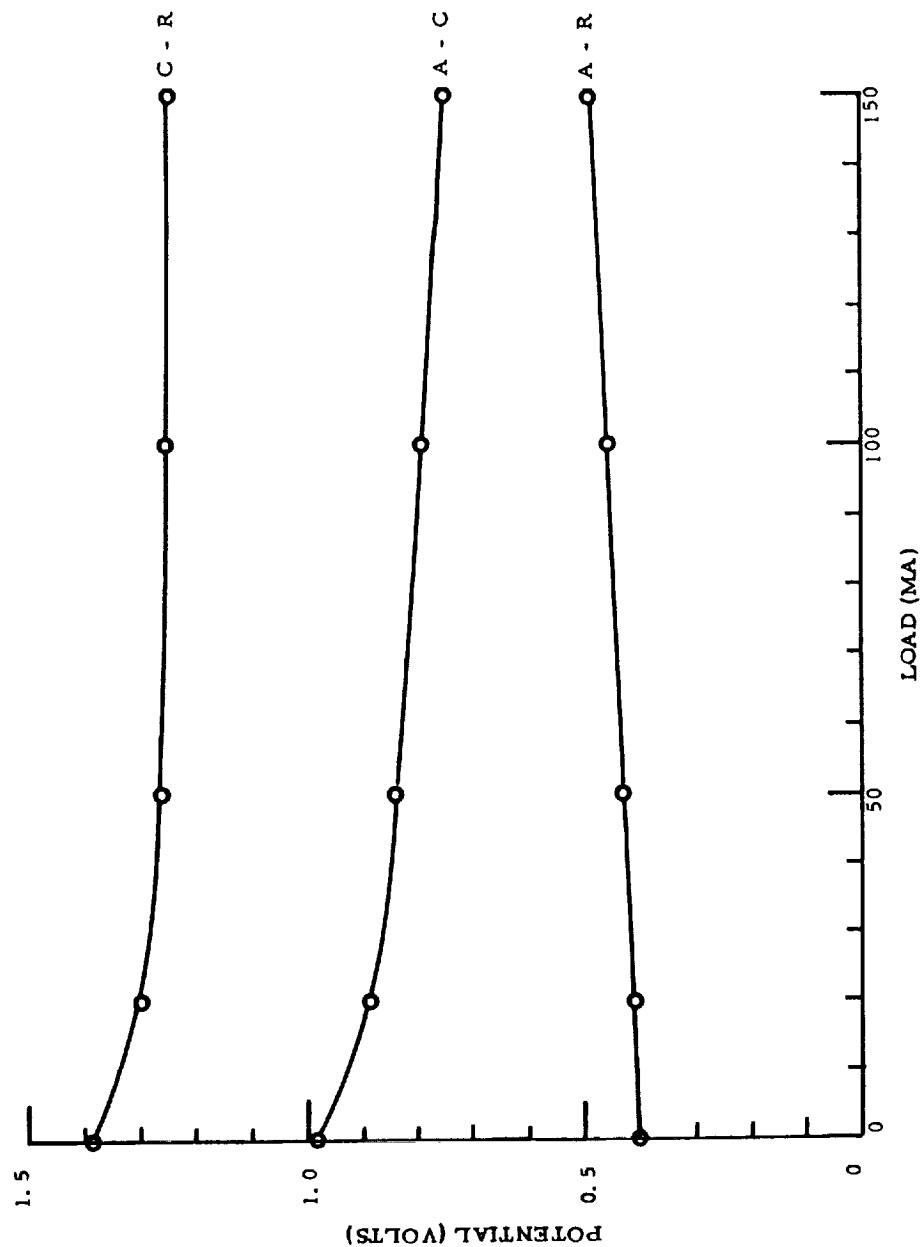


Time (2 sec/cm)

Fig. 6 Oscilloscope Trace of Anode-Cathode
Fluctuations at 220 Pulses/Min.

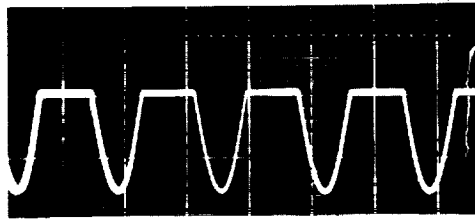
FIGURE 7

TABLE II DATA - POLARIZATION CURVES AFTER ELECTROLYTE PULSING



NOTE: A = Anode; C = Cathode; R = Reference

Potential
(200 mv/cm)

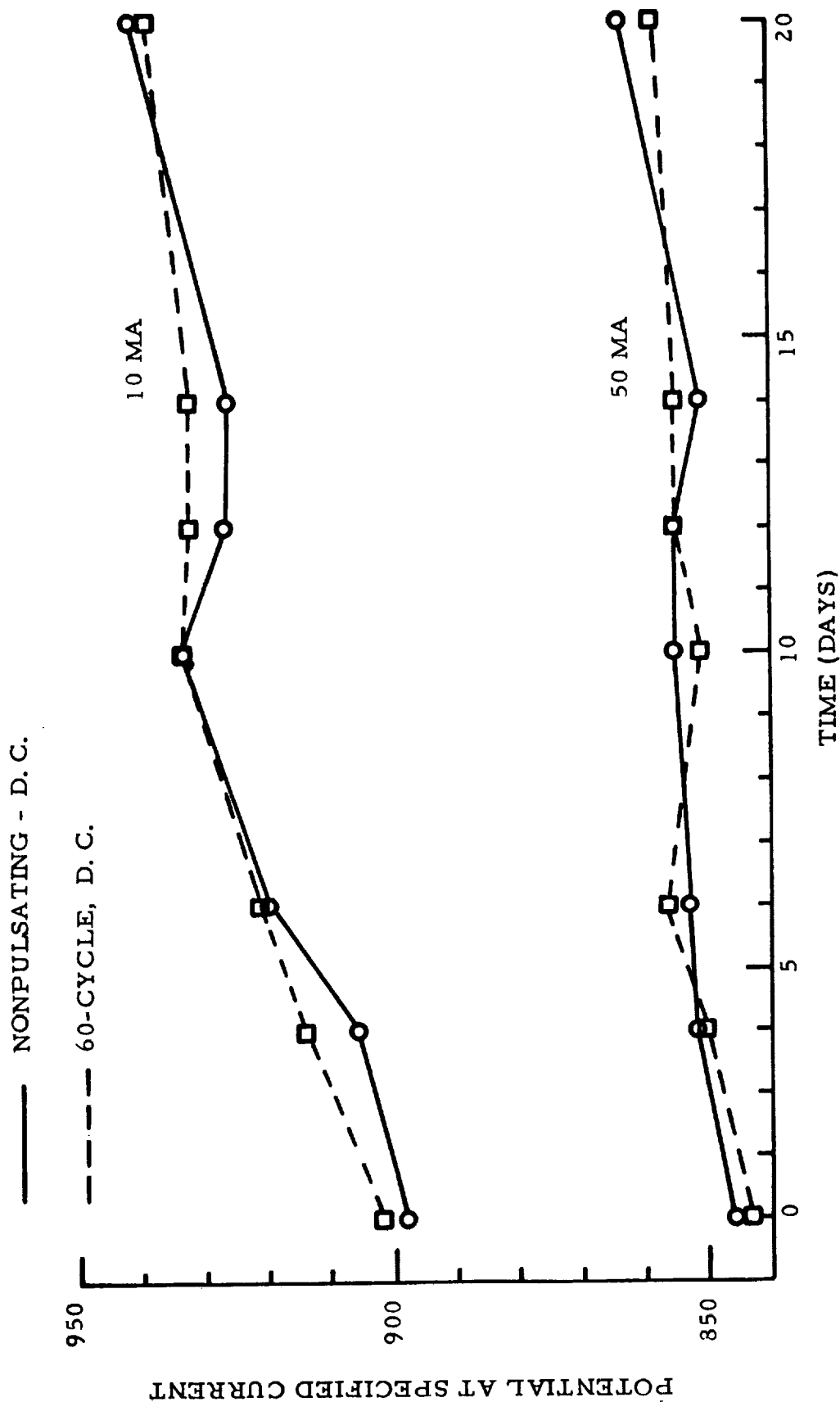


Time (10 millise. /cm)

Fig. 8 Oscilloscope Trace of Pulsating 60-Cycle
Direct Current - 50% Duty Cycle

FIGURE 9

TABLE III DATA - POLARIZATION CURVES FOR ELECTRICALLY-PULSED CELLS



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